An unprecedented cyclometallated platinum(II) complex incorporating a phosphinine co-ligand: synthesis and photoluminescence behaviour†

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The first luminescent cyclometallated platinum(II) complex bearing phosphinine co-ligands is reported. The long-lived luminescence, which it displays at 77 K, is assigned to a phosphinine-localised triplet excited state on the basis of electrochemical data and TD-DFT calculations.

Cyclometallated Pt(II) complexes have been increasingly studied over the last decade, with a view to applying their attractive luminescence properties in devices such as organic light-emitting diodes.1,2 These complexes typically comprise of a cyclometallating ligand – such as 2-phenylpyridine or related aryl-heterocycles – and various co-ligands completing the coordination sphere. Co-ligands frequently used include β-diketonates (e.g. acac)3, 2-picolinate,4 acetylides,5 and N-heterocyclic carbenes.6 In these cases, the lowest-energy excited states normally involve primarily the metal and the cyclometallated ligand, with the co-ligands playing only a minor role. Only in a few instances, such as 8-hydroxy-quinolinate, does the co-ligand participate significantly in the emissive excited state.7

Surprisingly, phosphorus based co-ligands remain relatively unexplored, despite having shown some promise,8 and we are unaware of any cyclometallated platinum(II) lumiphore with phosphinine co-ligands. Phosphinines – the phosphorus analogues of pyridines – constitute an intriguing family of ligands owing to their ability to act as strong π-acceptors. Since their discovery by Märtel et al.,9 investigations into their coordination chemistry have essentially been limited to low valence transition metal carbonyl complexes.10 The tedious procedures to synthesise phosphinines and their inadequate stability to harsh reaction conditions have hampered the exploration of their coordination chemistry with many transition metals.11 However we note that several important papers by Müller et al. have appeared recently dealing with the development of chelating phosphinine ligands in order prepare coordination compounds with transition metals in medium to high oxidation states such as Pd(II), Pt(II) and Ir(III).12 Given the high affinity of soft Pt(II) metal centres for phosphorus, it is surprising that there are so few reports on the coordination chemistry of phosphinines with Pt(II),13 and no studies into the electronic excited states of such compounds.

We were intrigued to explore the coordination chemistry of the {Pt(N′C-ppy)} moiety with the monodentate 2,6-diphenyl-4-methylphos-phinor (Lp) (Scheme 1), and to investigate the effect of such strongly π-acidic co-ligands on the excited-state properties. During the course of our work, an octahedral tris-P-C-cyclometallated iridium(III) complex was reported but the compound proved to be non-emissive even at low temperature, apparently due to efficient non-radiative decay processes.14 This is in stark contrast with the complex reported in this communication that exhibits a phosphinine based phosphorescence for the first time in a metal complex.

It proved possible to synthesise the target complex [Pt(ppy)(Lp)]CF3SO3 under very mild reaction conditions (Scheme 1). The first step involved chloride abstraction from compound [Pt(ppy)(μ-Cl)]2 using silver triflate in acetone. The resulting solvate species [Pt(ppy)(acetone)]CF3SO3 was treated directly with just over two equivalents of Lp. After 2 hours at room temperature, the reaction was worked up and the novel complex [Pt(ppy)(Lp)]CF3SO3 (complex 1) was isolated in high yield (see Supplementary Information for experimental details†). The mild reaction conditions employed are in stark contrast with the rather harsh synthetic protocols typically associated with reactions in which the chlorides are displaced,14 but are crucial in facilitating the introduction of the sensitive phosphinines into the coordination sphere.

Complex 1 was characterized by multinuclear NMR and IR spectroscopy, elemental analysis, and by a single crystal X-ray diffraction study.† The 31P{1H} NMR spectrum is especially informative (Fig. 1). It exhibits two doublets at 157.7 ppm and 133.3 ppm, attributed to the chemically different phosphinine centres, with 31Pt satellites of JP,p = 1760 Hz and JP,n,p = 4470 Hz respectively. Cis-P-P coupling (Jp,p = 46 Hz) is also observed, unambiguously confirming the coordination of two Lp ligands through phosphorus to the Pt(II) centre in solution. The inequivalence of the phosphinine ligands is also evident in the 1H NMR spectrum (Fig. S1†). 31P NMR and IR spectroscopy confirm the presence of the triflate anion.
The solid-state structure of this first member of a novel type of phosphinine complexes was unequivocally ascertained by a single crystal X-ray diffraction study. The molecular structure in the crystal (Fig. 2) clearly shows that complex 1 has a distorted square planar geometry. The Pt(II) centre is coordinated by the phosphorus atoms of two Lₚ ligands in a cis-fashion, and trans to the cyclometalated ppy ligand. Unfortunately, because of the serendipitous orientation of the ppy ligand, the positions of the nitrogen and cyclometalated carbon cannot be distinguished, prohibiting any deductions as to the Pt–N, Pt–P and Pt–C bond distances and angles.

The UV-visible absorption spectra of the phosphinine proligand Lₚ and of the Pt(II) complex 1 in dichloromethane solution at room temperature are shown in Fig. 3, with numerical data in Table 1. The spectrum of the proligand is dominated by an intense broad band centred at 278 nm (ε = 43700 M⁻¹ cm⁻¹), accompanied by a shoulder at 312 nm, and a weak, low-energy tail, around 350 nm. There have been very few studies of the photophysical properties of phosphinines, but we note that the spectrum is similar to that of the related 2,6-thienyl-4-methylphosphinine, albeit somewhat blue-shifted. The intense bands can reasonably be attributed to π→π* transitions in the aromatic rings; lower-energy transitions would be expected in the thienyl derivative owing to the more electron-donating nature of thiopephane compared to benzene. On the basis of TD DFT calculations, Mueller et al. assigned a similarly weak low-energy tail to a symmetry-forbidden Sₗ→S₀ transition.

The absorption spectrum of the Pt(II) complex 1 shows an intense band at a similar value to Lₚ, presumably due to the superposition of intraligand transitions in the phosphinine with those of the phenylpyridine, but there are clearly now additional, moderately intense bands in the 330–400 nm region (Fig. 3, Table 1). Bands in this region are typically found in Pt(N/C) complexes (where N/C represents a cyclometalated aryl-heterocycle such as phenylpyridine), due to charge-transfer transitions, normally from the metal/aryl ring to the heterocycle, according to theoretical calculations.

![Fig. 1](image)

**Table 1** Photophysical data for phosphinine Lₚ and complex 1

<table>
<thead>
<tr>
<th></th>
<th>λ_{max} / nm (ε / M⁻¹ cm⁻¹)</th>
<th>λ_{max}^{600} / nm</th>
<th>τ_{em} / µs</th>
</tr>
</thead>
<tbody>
<tr>
<td>Proligand Lₚ</td>
<td>278 (43700), 312sh (17900), 352sh (1350)</td>
<td>405(F), 582(P)</td>
<td>18.7</td>
</tr>
<tr>
<td>Complex 1</td>
<td>270 (54000), 332 (19300), 374sh (5220)</td>
<td>586</td>
<td>230</td>
</tr>
</tbody>
</table>

a Absorption data in CH₂Cl₂ at 298 K. b Emission data in diethyl ether / isopentane / ethanol (Lₚ) or CH₂Cl₂ (1) at 77 K; F and P indicate proposed assignment of fluorescence or phosphorescence to observed bands of Lₚ. c Fluorescence lifetime too short to be determined reliably.

At room temperature, no significant emission can be detected from the platinum complex 1. In contrast, intense emission is observed at 77 K (Fig. 3, Table 1). It is striking that the emission maximum under these conditions is very similar to that of the phosphorescence band displayed by Lₚ, but the lifetime is shorter by two orders of magnitude (τ_{em} = 230 µs) and there is no accompanying fluorescence band. These observations suggest that the emission in the complex emanates from an excited state that retains similar character to that of the free proligand Lₚ. Nevertheless, there must be sufficient influence of the metal through spin-orbit coupling to promote (i) Sₗ→T inter-system crossing, such that fluorescence from the singlet state is quenched, and (ii) T₁→S₀ phosphorescence, such that the radiative rate constant is increased and the lifetime concomitantly shortened. At room temperature,
non-radiative decay pathways are evidently able to compete effectively with the slow radiative decay process, quenching the emission. This behaviour contrasts with that displayed by the majority of Pt(ppy)(LL)\(^+\) and Pt(ppy)(LX) complexes, where the excited state is typically based on the Pt(ppy) unit, with enough metal character that spin-orbit coupling pathways increase the radiative rate constant to the extent that luminescence is observable at room temperature, typically with a lifetime of around a microsecond.\(^5\) Support for this interpretation is provided by TD DFT calculations on complex I, carried out as described in the Supporting Information.\(^6\) The \(T_1\) excited state is predominantly HOMO→LUMO in character. The density difference plots for the \(T_1\) excited state, calculated at the energy-minimised \(T_1\) geometry, are shown in Fig. 4. They confirm that the excited state is localised largely on the phosphinine ligand that is trans to the metallated carbon. The metal makes only a minor contribution to the excited state, such that a rather long luminescence lifetime would be anticipated, as observed.

Electrochemical measurements can sometimes also provide insight into the location of the frontier orbitals. In the present instance, cyclic voltammetry of I in solution in \(\text{CH}_2\text{Cl}_2\) reveals a well-defined, partially reversible reduction wave at \(-1.81\) V versus the ferrocene/ferrocenium couple, anodically shifted compared to \(E_{\text{p}}\) (\(E_{\text{p}}^{\text{ox}} = -2.45\) V). This contrasts with the behaviour of related complexes such as \([\text{Pt}(\text{ppy})(\text{acac})]\), where reduction is not observed within the solvent accessible window, supporting the notion that the LUMO lies on the phosphinine. On the other hand, no clearly defined oxidations could be observed.

In summary, the first example of a cyclometallated platinum(II) complex incorporating phosphinine co-ligands has been described, in which the ligands are bound through phosphorus to a cyclometallated platinum centre. The complex is strongly luminescent at low temperature, from an excited state associated primarily with one of the phosphinine ligands. The study may open up further opportunities for introducing phosphinine ligands into novel metal complexes that display interesting physical properties.

Notes and references

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‡ Electronic Supplementary Information (ESI) available: details of experimental procedure, NMR spectra, density functional theory calculations, and instrumentation for photophysical and electrochemical measurements. A CIF file containing crystallographic data of complex I is available at CCDC 982683. See DOI: 10.1039/b000000x/

\(^6\) The spectrum was independent of concentration over an order of magnitude range, ruling out the possibility that the low-energy band is due to aggregation.


Crystal data for 1. yellow block-like crystals: C_{51}H_{44}F_{3}NO_{4}P_{2}PtS, monoclinic, C 2/c, a = 19.8832(5), b = 20.2082(5), c = 14.8191(4) Å, \( \beta = 128.153(1) \) \(^\circ\), V = 4682.3(2) Å\(^3\), Z = 4, T = 200(2) K, \( \mu = 3.167 \) mm\(^{-1}\), 23707 reflections measured, 6810 independent (R\(_{int}\) = 0.0176), 6391 observed \([I\geq 2\sigma(I)]\), 278 parameters, final R indices R\(_1\) [I\(\geq 2\sigma(I)\)] = 0.0253 and wR\(_2\) (all data) = 0.0721, GOF on F\(^2\) = 1.103, max/min residual electron density = 1.88/ -0.71 e.Å\(^{-3}\). CCDC 982683.