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1 **Ground air: a first approximation of the Earth's second largest reservoir of**  
2 **carbon dioxide gas**

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6

7 **It is becoming increasingly clear that a substantial reservoir of carbon exists in the**  
8 **unsaturated zone of aquifers, though the total size of this reservoir on a global scale**  
9 **remains unquantified. Here we provide the first broad estimate of the amount of**  
10 **carbon dioxide gas found in this terrestrial reservoir. We calculate that between 2 and**  
11 **53 PgC exists as gaseous CO<sub>2</sub> in aquifers worldwide, generated by the slow microbial**  
12 **oxidation of organic particles transported into aquifers by percolating groundwater.**  
13 **Importantly, this carbon reservoir is in the form of CO<sub>2</sub> gas, and is therefore**  
14 **transferable to the Earth's atmosphere without any phase change. On a coarse scale,**  
15 **water table depths are partially controlled by local sea level; sea level lowering**  
16 **therefore allows slow carbon sequestration into the reservoir and sea level increases**  
17 **force rapid CO<sub>2</sub> outgassing from this reservoir. High-resolution cave air pCO<sub>2</sub> data**  
18 **demonstrate that sea level variability does affect CO<sub>2</sub> outgassing rates from the**  
19 **unsaturated zone, and that the CO<sub>2</sub> outgassing due to sea level rise currently occurs**  
20 **on daily (tidal) timescales. We suggest that global mean water table depth must**  
21 **modulate the global unsaturated zone volume and the size of this carbon reservoir,**  
22 **potentially affecting atmospheric CO<sub>2</sub> on geological timescales.**

23 **Keywords: caves, carbon reservoirs, ground air, carbon dioxide, vadose zone**

24 **1. Introduction**

25 The presence of a reservoir of carbon within the unsaturated zone of karst aquifers is now  
26 well-established (e.g., Matthey et al., 2016; Noronha et al., 2015). Calculations based on

27 groundwater geochemistry have long suggested that groundwater may equilibrate with air  
28 that has  $p\text{CO}_2$  substantially higher than soil air  $p\text{CO}_2$ , implying a deeper source (Atkinson,  
29 1977). This early research concluded that microbial oxidation of mechanically transported  
30 organic material within aquifer permeability generates  $\text{CO}_2$  within the unsaturated zone of  
31 karst aquifers (Atkinson, 1977; Wood, 1985). This air reservoir, termed 'ground air', is  
32 characterised by very high  $\text{CO}_2$  concentrations. Recent studies from cave sites (e.g., Baldini  
33 et al., 2006; Bourges et al., 2001; Whitaker et al., 2010) have shown that cave air  $p\text{CO}_2$   
34 values generally increase in smaller or more sheltered passages, with values sometimes  
35 considerably higher than local soil  $p\text{CO}_2$ , suggesting the presence of ground air. However,  
36 direct measurements of borehole air confirm that a reservoir of extremely high  $p\text{CO}_2$  air  
37 exists within the unsaturated zone of aquifers in a variety of different lithologies (Benavente  
38 et al., 2010; Hendry et al., 1993; Hendry and Wassenaar, 2005), not just karstic aquifers. For  
39 example, research on the gas content of siliciclastic deposits of the Ogalla aquifer in south  
40 Texas concluded that aerobic microbes oxidized organic carbon transported to intergranular  
41 porosity by recharge water, producing  $\text{CO}_2$  (Wood and Petraitis, 1984). Furthermore,  
42 radiocarbon measurements support the concept that this  $\text{CO}_2$  is derived from the decay of  
43 old carbon that was probably transported into the aquifer (Bergel et al., 2017; Lechleitner et  
44 al., 2016; Noronha et al., 2015; Wood et al., 2014), rather than  $\text{CO}_2$  produced in the soil  
45 zone and then diffused downward. Considered together, existing geochemical evidence  
46 suggests the presence of a substantial carbon dioxide reservoir at depth that has largely  
47 escaped quantification. Access issues have meant that this reservoir is most easily identified  
48 in cavernous and karstified environments, but ground air is found in any lithology with even  
49 small-scale permeability.

50

51 Here we use new laboratory and field data combined with published estimates of mean  
52 global depth to groundwater (Fan et al., 2013; Serrano-Ortiz et al., 2010) to estimate the size  
53 of the global ground air carbon reservoir. A strong link between sea level and  $\text{CO}_2$   
54 outgassing from this reservoir is observed in a new cave air  $p\text{CO}_2$  dataset, which implies that

55 vertical groundwater shifts associated with local sea level push ground air out of the  
56 subsurface during a rising tide and pull atmospheric air into the subsurface during a falling  
57 tide (i.e., that the water table acts as a piston). We suggest that eustatic sea level increases  
58 on geological timescales potentially also forced CO<sub>2</sub> out of the ground air reservoir and into  
59 the atmospheric reservoir, potentially accounting for a portion of the observed atmospheric  
60 CO<sub>2</sub> increase.

61

62

## 63 **2. Methods**

64

### 65 **2.1. Quantifying the global ground air reservoir**

66 To constrain the global ground air CO<sub>2</sub> reservoir we estimated: *i*) mean ground air  $p\text{CO}_2$ , *ii*)  
67 mean global depth to groundwater, *iii*) mean global net (primary and secondary) permeability  
68 of the unsaturated zone, and *iv*) global land surface area (Table 1). Considerable variability  
69 exists in all these parameters, and we therefore necessarily report a broad range of ground  
70 air carbon reservoir sizes. We believe that the true value lies within this range, and future  
71 studies should focus on better constraining the variables defining this range.

72

73 The current global unsaturated zone volume was estimated here using published exposed  
74 land area and the Global Mean Water Table Depth (GMWTD: the mean distance from the  
75 surface to the water table over all points on land) values; the minimum GMWTD (26 m) was  
76 calculated using data presented in Fan *et al.* (2013) and the maximum (100 m) uses the  
77 value reported in Serrano-Ortiz *et al.* (2010)). The land area estimate does not account for  
78 ice cover because of substantial uncertainties in both the amount and distribution of  
79 subglacial carbon. Bedrock permeability values range considerably (Freeze and Cherry,  
80 1979). We use a conservative mean global value of 10%, consistent with previous estimates  
81 (Serrano-Ortiz *et al.*, 2010). This includes both primary and secondary permeability, and  
82 accounts for decreasing permeability with depth (Williams, 2008). Ground air  $p\text{CO}_2$  values

83 were assumed to range from 12,000 ppmv to 70,000 ppmv based on the results of the  
84 laboratory and field experiments conducted here and published data (see Supplementary  
85 Content). The minimum value is probably very conservative, but given the substantial  
86 uncertainties involved in this first estimate of the global ground air carbon reservoir size we  
87 feel that the broad range of estimates is justified.

88

## 89 **2.2. Cave air $p\text{CO}_2$ monitoring**

90 Cave air  $p\text{CO}_2$  measurements were made in Conch Bar Caves, Middle Caicos, Turks and  
91 Caicos Islands (21°49'34"N, 71°47'28"W) to gauge the response of ground air to local sea  
92 level fluctuations. The cave is a flank margin cave, developed in Cretaceous and Tertiary  
93 aged carbonate platform sediments. The cave has numerous entrances, is well ventilated,  
94 and has a number of saltwater pools fed by direct connections to the sea (Supplementary  
95 Figure 3) (Smart et al., 1992).

96

97 The  $p\text{CO}_2$  logger was placed in a small cave chamber with good airflow 180 meters from the  
98 nearest entrance, 20 meters below the surface, and two meters above mean sea level  
99 (Supplementary Figure 3). The majority of the chamber floor was flooded during high tide,  
100 except for a few isolated 'islands' of bedrock or secondary calcite (< four meters in diameter)  
101 that remained above sea level. The  $p\text{CO}_2$  logger was placed on one of these, and was  
102 always at least one meter above the water level in the chamber. Cave air  $p\text{CO}_2$  was  
103 measured automatically every three hours for 318 days from April 17, 2011, to February 28,  
104 2012, using a calibrated Vaisala GMP343 infrared carbon dioxide probe connected to a  
105 Vaisala MI70 indicator ( $\pm 7$  ppmv) (Ridley et al., 2015). Data were corrected for barometric  
106 pressure (also measured on site, using a Barotroll barometric pressure logger) using the  
107 method outlined in Spötl *et al.* (2005). Spectral analysis of the  $p\text{CO}_2$  dataset was conducted  
108 using PAST software (Hammer et al., 2001).

109

## 110 **2.3. Beach transect $\text{CO}_2$ measurements**

111 Measurements of ground air in the unsaturated zone were made at five sandy beaches  
112 across the UK (July-September 2013). Each site was divided into three zones (intertidal,  
113 high beach, and dune) and measurements were taken within each. Measurements were  
114 made across transects orthogonal to the shoreline, crossing all three zones at each of the  
115 five beach sites. At each location, a calibrated Vaisala GMP343 combination CO<sub>2</sub> and  
116 temperature probe (uncertainties of ±0.04% for CO<sub>2</sub> and ±0.05°C for temperature) was  
117 buried to a depth of one meter (or to just above the water table if the water table was  
118 shallower than one meter). CO<sub>2</sub> values stabilised at all sites within 100 minutes. The  
119 intertidal zone represents an area where the ground air CO<sub>2</sub> signature is 'reset' to  
120 atmospheric values once the tide recedes from the zone and atmospheric air is drawn into  
121 the subsurface by the dropping sea level. The sea occasionally affects the high beach zone  
122 environment during storms and unusually high tides, but not on daily timescales. The dune  
123 zone was not submerged in the recent past, and is overlain by typical halophytic vegetation  
124 and by a thin (< 5 cm), immature soil zone consisting almost exclusively of an O-horizon  
125 directly above the quartz sand substrate. The dune zone provides a contrast to the other  
126 two zones due to the presence of soil organic material, and because there would have been  
127 sufficient time for the organic material to infiltrate the sand substrate and oxidise. The dunes  
128 thereby provide an environment where ground air in the unsaturated zone is reasonably  
129 accessible. Time-series monitoring was conducted in the dune environment of Camber  
130 Sands and Greatstone Beaches, Kent, UK, where data was logged automatically every 15  
131 minutes over several days.

132

### 133 **3. Results and Discussion**

134

#### 135 **3.1. Existing evidence for 'ground air'**

136 The evidence for air within the vadose zone with substantially elevated CO<sub>2</sub> (and methane)  
137 concentrations is now strong. Laboratory mesocosm experiments (Hendry et al., 1993;

138 Hendry et al., 2001), new field data, and previously published data (e.g., Atkinson, 1977;  
139 Batiot-Guilhe et al., 2007; Denis et al., 2005; James, 1977; Matthey et al., 2016; Serrano-Ortiz  
140 et al., 2010; Wood and Petraitis, 1984) collectively indicate that a substantial CO<sub>2</sub> pool exists  
141 in the unsaturated zone of aquifers worldwide. Previous researchers have even suggested  
142 that the majority of cave air CO<sub>2</sub> is sourced from a deep biogenic source (Breecker et al.,  
143 2012), rather than the soil. We have compiled a representative collection of published  
144 measurements of unsaturated zone air  $p\text{CO}_2$  and  $\delta^{13}\text{C}$  (based on 14 different sites from  
145 different environments), which strongly suggest variable mixing between two end member  
146 pools of CO<sub>2</sub>: one with low  $p\text{CO}_2$  and high  $\delta^{13}\text{C}$  and a second with substantially elevated  
147  $p\text{CO}_2$  and low (but locally variable)  $\delta^{13}\text{C}$  values. The first pool is clearly the Earth's  
148 atmosphere, whereas the second represents a reservoir with  $p\text{CO}_2$  that is up to two orders  
149 of magnitude higher than typical soil air  $p\text{CO}_2$  (Murthy et al., 2003) (Figure 1). Because most  
150 of these elevated measurements are from regions with no known magmatic or hydrocarbon  
151 related CO<sub>2</sub>, this strongly supports previous studies concluding that high CO<sub>2</sub> ground air  
152 exists in the unsaturated zone. Furthermore, if the second reservoir were simply soil air,  
153 mixing would reflect the photosynthetic pathway of the vegetation overlying the various sites  
154 (e.g., between -22 and -25‰ VPDB for C<sub>3</sub> vegetation and between -10 and -15 VPDB for C<sub>4</sub>  
155 vegetation). However, average mixing lines indicate that the CO<sub>2</sub> reservoir typically has a  
156  $\delta^{13}\text{C}$  of between -17 and -19‰ VPDB (although some individual sites clearly do reflect  
157 modern overlying vegetation, such as Obir Cave), suggesting that soil is not the main source  
158 of the CO<sub>2</sub>. Possible sources for CO<sub>2</sub> found at depth in non-geothermal areas include:  
159 diffusion from the soil zone, microbial oxidation of organic material at depth (either material  
160 transported downward from the soil or carbon deposited with the rock), or degassing during  
161 calcite precipitation at the surface of the water table. The observed carbon isotope ratios  
162 may reflect mixing of organic material filtered by the aquifer over thousands of years, thereby  
163 integrating the  $\delta^{13}\text{C}$  signal of a variety of vegetation, sometimes averaging C<sub>3</sub> and C<sub>4</sub>  
164 vegetation signatures. This is strongly supported by radiocarbon evidence from stalagmites  
165 and cave air suggesting the contribution of substantial amounts of very old carbon, and that

166 soil carbon is often not the direct source of cave air  $p\text{CO}_2$  (Noronha et al., 2015). The recent  
167 use of oxidative ratios of subsurface gases provides more strong support for the concept that  
168 the carbon in both caves and the vadose zone is at least centuries old (Bergel et al., 2017).  
169 Furthermore, studies on dissolved organic carbon within an aquitard demonstrate that C  
170 within connate pore water is approximately 15,000 years old (Hendry and Wassenaar,  
171 2005). The high (compared to  $\text{C}_3$  vegetation)  $\delta^{13}\text{C}$  values typical of ground air may also  
172 reflect carbonate equilibrium chemical reactions involving both bedrock dissolution and  
173 calcite precipitation at the water table. Matthey et al. (2016) provide a comprehensive review  
174 of ground air in karstic environments and how advective and diffusive mixing of  $\text{CO}_2$  derived  
175 from different sources, including soil air, occurs.

176

177 The natural environments with the highest ground air  $p\text{CO}_2$  values are: *i*) inaccessible  
178 small-scale permeability within bedrock and *ii*) deep, unventilated cave and mine passages,  
179 which are inaccessible without breathing apparatus due to the high  $p\text{CO}_2$  levels (known  
180 colloquially as 'bad' or 'foul' air amongst cavers (Smith, 1999)). Cave air  $p\text{CO}_2$   
181 measurements made in more accessible sections of caves reflect, almost without exception,  
182 a mixture of ground air with substantial amounts of outside (atmospheric) air and have  $p\text{CO}_2$   
183 values low enough to permit exploration of the passage. In one of the few examples from a  
184 poorly ventilated passage (in Lascaux Cave, France), Peyraube *et al.* (2013) measured  
185  $p\text{CO}_2$  values over 70,000 ppmv. Additionally, a growing number of borehole  $p\text{CO}_2$   
186 measurements (Affek et al., 1998; Benavente et al., 2010; Peyraube et al., 2013; Vadillo et  
187 al., 2010) with maximum values approaching 70,000 ppmv also indicate that ground air is  
188 present. It is intriguing that almost no measurements of ground air considerably above  
189 70,000 ppmv exist. The reasons underlying this observation are unclear, but may reflect a  
190 reduction in metabolic rate of aerobic bacteria (and associated organic matter oxidation rate)  
191 once ground air oxygen levels drop below 14% (equivalent to the conversion of 70,000 ppmv  
192  $\text{O}_2$  gas to  $\text{CO}_2$  gas from the presumed initial  $p\text{O}_2$  value of 210,000 ppmv (21%, the  
193 concentration in the Earth's atmosphere)).



194

195 Available data suggest that ground air  $p\text{CO}_2$  values are greatest near the capillary fringe and  
196 decrease upward towards the soil zone (Wood et al., 2014). This is due to enhanced  $\text{CO}_2$   
197 generation near the water table but also to dissolution and downward transport of  $\text{CO}_2$  by  
198 percolation waters (Affek et al., 1998; Walvoord et al., 2005; Wood et al., 2014). Calcite  
199 precipitation at the water table could also partially account for the high concentrations  
200 adjacent to the water table, but mass balance considerations suggest that microbial  
201 oxidation of organic matter is a larger source (Walvoord et al., 2005). At some borehole  
202 sites, it is clear that high permeability, even without the presence of cavernous porosity,  
203 creates conditions favouring rapid air exchange between the surface and subsurface, and  
204 the residence time of vadose zone air is measurable in years to decades (Thorstenson et al.,  
205 1998). In these cases, such as at Yucca Mountain, Nevada, ground air  $p\text{CO}_2$  values are  
206 moderated by exchange with the atmosphere, with very low values, typically ranging from  
207 900 to 6,000 ppmv according to local permeability and depth (Thorstenson et al., 1998). This  
208 illustrates that ground air  $p\text{CO}_2$  varies substantially both geographically and vertically. A  
209 number of different variables, including bedrock permeability, moisture content, rock type  
210 and organic content, local climate, and vegetation cover all affect ground air  $p\text{CO}_2$  values.  
211 The maximum measured values for ground air  $p\text{CO}_2$  used here (70,000 ppmv) are therefore  
212 likely substantially higher than mean values of the global unsaturated zone reservoir.  
213 Conversely, the minimum values (12,000 ppmv, derived from our lab and field data (see  
214 Supplementary Content)) are lower than most direct ground air  $p\text{CO}_2$  measurements, and  
215 therefore are likely to underestimate the global mean.

216

### 217 **3.2. Estimating the ground air reservoir size**

218 Serrano-Ortiz *et al.* (2010) estimated the  $\text{CO}_2$  contained in karstic regions, but did not  
219 consider non-karstic areas. However, no reason exists why only karst regions should host  
220 ground air and, as discussed previously, unsaturated zone  $p\text{CO}_2$  measurements in other  
221 environments support ground air as a global phenomenon. We therefore suggest that

222 elevated  $p\text{CO}_2$  exists throughout the unsaturated zone globally, but that direct  
223 measurements are lacking due to the absence of accessible cave passage in non-limestone  
224 lithologies. In fact, non-karstic aquifers may actually have higher mean ground air  $p\text{CO}_2$   
225 values due to a reduced capacity for ventilation due to the absence of large passages  
226 (Covington, 2016). The presence of aerobic bacteria in the deep subsurface is currently not  
227 well constrained, but a number of studies now illustrate that aerobic bacteria are found in  
228 some of the harshest and least hospitable environments on the planet, including in ocean  
229 sediment at depth in the most nutrient-poor regions of the Pacific Ocean (D'Hondt et al.,  
230 2015), within caves (e.g., Tomova et al., 2013), and in bedrock (Personne et al., 2004).  
231 Specifically, studies on boreholes demonstrate that aerobic bacteria exist throughout  
232 subsurface and that their concentrations do not seem to decrease with depth (Hicks and  
233 Fredrickson, 1989). The current consensus appears to favour a model where the biosphere  
234 in the deep subsurface is both diverse and active (Fredrickson and Balkwill, 2006; McMahon  
235 and Parnell, 2014; Rempfert et al., 2017), so the presence at depth of microbes capable of  
236 oxidising organic matter is not surprising.

237

238 Using the estimated ranges in parameters affecting global unsaturated zone volume and  
239 ground air  $p\text{CO}_2$  (Table 1), we estimate between 2 and 53 petagrams of carbon (PgC) are  
240 stored as ground air within the unsaturated zone of aquifers globally. This range is  
241 consistent with values of 2.0 PgC calculated by Serrano-Ortiz *et al.* (2010) calculated for just  
242 karst regions (representing ~15% of land area) using lower values of  $p\text{CO}_2$  measured in  
243 caves (20,000 ppmv). This range is also consistent with the calculations suggesting 10-100  
244 PgC exists in the deep biosphere as microbes (between 2 and 19% of the Earth's total  
245 biomass) (McMahon and Parnell, 2014; Whitman et al., 1998). Ground air  $\text{CO}_2$  therefore  
246 represents a terrestrial C pool containing between 0.24 and 6.4% of the current atmospheric  
247 C content (830 PgC) (Le Quere et al., 2015). The calculation is most sensitive to the  
248 GMWTD, and simply changing the value from the Serrano-Ortiz et al. (2010) value (0.1 km)  
249 to the Fan et al. (2013) value (0.026 km) reduces the maximum ground air reservoir value

250 from 53 PgC to 14 PgC. The lower estimate of GMWTD of Fan et al. (2013) is more  
251 comprehensive, and consequently it is likely that the total ground air reservoir is on the lower  
252 end of the range reported here. Critically however, unlike many other non-atmospheric  
253 carbon reservoirs, ground air C exists as gaseous CO<sub>2</sub>, and does not require a phase  
254 change prior to entering the atmospheric pool. For example, carbon stored in the deep  
255 marine reservoir has a mean residence time of ~100,000 years, while carbon in limestone  
256 has a residence time of ~100 million years. Carbon within the biosphere is more mobile  
257 (mean residence time of living terrestrial biosphere = ~20 years), but with the exception of  
258 fires is not instantaneous. Consequently, variability in unsaturated zone reservoir magnitude  
259 could affect atmospheric CO<sub>2</sub> concentrations and ultimately global climate extremely quickly.  
260 Any major rise in sea level would necessarily be accompanied by ground air outgassing that  
261 reflects the rapidity of the sea level change.

262

263

### 264 **3.3. Correlations between cave air pCO<sub>2</sub> and sea level**

265 Evidence that even small tidal sea level fluctuations push high-pCO<sub>2</sub> air out of the  
266 unsaturated zone ground air reservoir and into cave passage is derived from new high-  
267 resolution pCO<sub>2</sub> time-series data from Conch Bar Caves (Turks and Caicos Islands)  
268 proximal to the Atlantic Ocean (Figure 3). Importantly, cave air pCO<sub>2</sub> increases with  
269 increasing local sea level, indicating that CO<sub>2</sub> is forced up from the bedrock permeability  
270 rather than down from the soil. The outgassing signature is remarkably clear despite the  
271 cave system having multiple entrances and an active ventilation system (Figure S3). The  
272 results are striking, with cave air pCO<sub>2</sub> tracking sea level, illustrating the ground air CO<sub>2</sub>  
273 'piston effect' well. Spectral analysis of both the Conch Bar Cave pCO<sub>2</sub> record and the tide  
274 gauge-derived sea level record illustrate in-phase 12- and 24-hour cycles (Figure 3),  
275 reflecting lunar tidal forces.

276

277 These observations have implications on longer timescales. GMSL reductions associated  
278 with low sea levels on geologic timescales (e.g., Ice Ages or glaciations) expose new land  
279 while simultaneously increasing GMWTD, thereby increasing the unsaturated zone volume.  
280 In most situations with unconfined aquifers, sea level acts as the local base level, and shifts  
281 in base level control the elevation of the water table further inland in accordance with the  
282 Dupuit equation (Fetter, 1994; Hiscock, 2005). In unconfined aquifers, basic hydrological  
283 principles dictate that water must flow from high hydraulic head to low hydraulic head; an  
284 increase in sea level is therefore propagated inland until it eventually affects the entire  
285 aquifer. Evidence does exist for sea level-induced water table lowering during periods when  
286 sea level was substantially lower. For example, substantial cave development ~100m lower  
287 than the modern water table in Florida may reflect local water table responding linearly to  
288 sea level rise during the last glacial termination (Wilson, 1988). In fact, cave development  
289 within the Floridan aquifer may reflect mixing of the fresh water table with high  $p\text{CO}_2$  ground  
290 air during the LGM (Gulley et al., 2013). Abundant evidence for a lower water table during  
291 glacial conditions exists throughout coastal regions globally (e.g., Bard et al., 2002; Moseley  
292 et al., 2013). The simple assumption of unconfined flow is not directly applicable to some  
293 groundwater basins with complex geological structural controls (such as the Basin and  
294 Range province of North America), but is relevant in many cases. We further acknowledge  
295 that shifts in climate and regional recharge conditions on long timescales also impact the  
296 depth to the water table locally, but globally these shifts would tend to cancel each other out  
297 (i.e., shifting rainfall patterns will raise the water table in one area while lowering it in  
298 another).

299

300 Downward percolating water will transport organic matter into the newly exposed volume of  
301 rock (or sediment) where oxidation produces  $\text{CO}_2$ . In this manner, sequestration of  
302 atmospheric  $\text{CO}_2$  will occur with sea level falls. On the other hand, sea level increases will  
303 cause flooding of land, reduced GMWTD, and a smaller ground air carbon reservoir (during  
304 low-ice volume intervals of Earth history). During transitions from high to low ice volume

305 intervals, some CO<sub>2</sub> gas will necessarily transfer from the unsaturated zone into the  
306 atmosphere. Interestingly, the identification of this terrestrial carbon reservoir is consistent  
307 with recent results suggesting increased storage of carbon during the Last Glacial (~21,000  
308 years before present) in an previously unidentified inert terrestrial pool (Ciais et al., 2012),  
309 which was apparently released into the atmosphere during deglaciation.

310

311

## 312 **5. Conclusions**

313 Here we calculate that between 2 and 53 PgC exist in a terrestrial carbon reservoir located  
314 in the unsaturated zone of aquifers worldwide. This range is consistent with previous  
315 estimates of carbon dioxide content of karst aquifers alone and with estimates of microbial  
316 biomass within all aquifers. We agree with the recently expressed perspective (Bergel et al.,  
317 2017) that the increasingly clear presence of a 'ground air' reservoir may require a re-  
318 evaluation of the classic models of carbon dioxide formation within karst aquifers.

319 Additionally, we propose that this reservoir is not restricted to karst aquifers, but is instead  
320 commonplace in all lithologies with any appreciable permeability. This global 'ground air'  
321 carbon reservoir is the second largest store of CO<sub>2</sub> gas on the planet, but remains largely  
322 unappreciated due to difficulties with access.

323

324 Assuming that the PCO<sub>2</sub> of ground air is (on average) temporally constant, the largest  
325 control on the carbon amount stored is the volume of the unsaturated zone. Variability in  
326 ground air carbon reservoir size represents a potential control on global atmospheric pCO<sub>2</sub>  
327 and consequently temperature. A new cave air PCO<sub>2</sub> dataset from a coastal cave illustrates  
328 that sea level is a fundamental control on the outgassing of CO<sub>2</sub> from the ground air  
329 reservoir. Changes in eustatic sea level will therefore directly influence the unsaturated zone  
330 volume and hence the global ground air reservoir, potentially affecting the amount of CO<sub>2</sub>  
331 contained within the Earth's atmosphere and, consequently, climate. It is worth noting that  
332 this mechanism may also have contributed to more pronounced climate shifts during

333 geological intervals when continental shelves were larger, or expansive shallow seas were  
334 present, such as the late Neoproterozoic or the Ordovician. In these cases, a moderate sea  
335 level drop would have exposed considerable amounts of land, possibly resulting in  
336 considerable carbon storage in the unsaturated zone followed by substantial release of CO<sub>2</sub>  
337 during sea level rises.

338

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500

501 **Figure Captions:**

502 **Figure 1: Keeling plot of published CO<sub>2</sub> measurements (in % atm) from the**  
503 **unsaturated zone and outside atmosphere.** Cave air, soil air, atmospheric air, well air,  
504 and borehole air pCO<sub>2</sub> data from selected sites from around the world (Batiot-Guilhe et al.,  
505 2007; Benavente et al., 2011; Bourges et al., 2001; Breecker et al., 2012; Denis et al., 2005;  
506 Frisia et al., 2011; Kowalczyk and Froelich, 2010; Matthey et al., 2010; Peyraube et al., 2012;  
507 Peyraube et al., 2013; Riechelmann et al., 2011; Spötl et al., 2005; Tremaine et al., 2011).  
508 Sites were chosen to illustrate ground air CO<sub>2</sub> in different environments, and are not  
509 comprehensive. Measurements from sites currently overlain by C<sub>4</sub> vegetation are  
510 represented by triangles and those overlain by C<sub>3</sub> vegetation by circles. Atmospheric values  
511 are for Mauna Loa Observatory (Keeling et al., 2001) (dark blue stars) and published values  
512 above some cave sites (light blue stars).

513

514 **Figure 2: Carbon dioxide concentrations along transects perpendicular to the coast at**  
515 **five beach locations in the UK.** Measurements of ground air in the unsaturated zones at  
516 five sandy beaches across the UK taken at different times between July-September 2013.  
517 Measurements were made across transects orthogonal to the shoreline, using a calibrated  
518 Vaisala GMP343 combination CO<sub>2</sub> and temperature probe buried to a depth of one meter (or  
519 to just above the water table if the water table was shallower than one meter). Two transects  
520 were conducted at Camber Sands on different days; these are labelled A and B respectively.

521

522 **Figure 3: Cave air record at Conch Bar Cave, Turks and Caicos Islands, compared**  
523 **with tide data.** (a) Cave air pCO<sub>2</sub> was measured from April 17, 2011, to February 28, 2012.  
524 One representative week (8-14 August 2011) of the cave air pCO<sub>2</sub> record is shown here,  
525 along with sea level data from the nearest tide gauge (Virginia Key, Florida, USA; 950 km to  
526 the NW). The time difference between the tide at Virginia Key and that measured live (no

527 logged data available) at Sandy Point, Turks and Caicos Islands, is less than one hour. (b)  
528 Spectral analysis of the full 318-day datasets (inset) illustrates the presence of statistically  
529 significant (at 90% confidence) 12-hour and 24-hour cycles within both the cave air pCO<sub>2</sub>  
530 and sea level datasets.

531

532 **Table 1: Parameters used to estimate ground air carbon stores.** The range of values  
533 used represents the uncertainty in the measurements. Land area values and global mean  
534 water table depth have varied over geological time; estimates are presented for the periods  
535 considered in this study.